

Bohr
↓
Wheeler
↓
Misner
↓
Ruitz
↓
you

I1. Quantization 1900

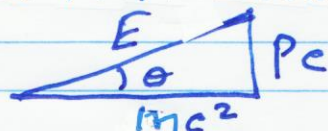
1. Energy Max Planck $E_n = nhf$ $n = 0, 1, 2, \dots$
Harmonic Oscillator The QM result is $E_n = (n + \frac{1}{2})hf$
2. Light Einstein 1905
 $E_{\text{photon}} = hf$
 ΔE_n same
 $\Delta E_n = hf$
3. Angular Momentum Bohr 1913
No radiation unless electron drops to lower energy level.
 $L = pr = mvr$
 $L = n \frac{h}{2\pi} \equiv n\hbar$



Gets you the Balmer Series and more.

I2. Matter Waves

Classical EM \Rightarrow shaking charge spiral in losing energy. Atoms should be unstable. Matter



For light $E = pc$

all energy is in pc max speed $\theta = 90^\circ$

$E = pc = hf$
 λf

$p\lambda = h$

$\lambda = \frac{h}{p}$

de Broglie \rightarrow true also for matter

You should not exist Bohr model

p	n	e ⁻
1913	1932	1897
particles		

Waves γ

1905 Photon \leftarrow pure energy.

Two Big Surprises

1. Light is particle also
2. Particles are waves also

I3. Wave Properties

$\psi(x,t) = A \sin[k(x-vt)]$

need for units of length to cancel Wavelength

Let $t=0$ $\psi(x,0) = A \sin kx$ $k\lambda = 2\pi$

$k = \frac{2\pi}{\lambda}$ Wave number

Similarly $k v T = 2\pi$ $k v = \frac{2\pi}{T} = 2\pi f = \omega$

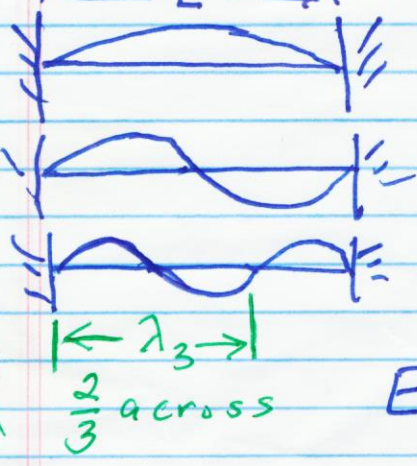
Period \rightarrow

Sometimes you see $\sin(kx - \omega t)$

angular velocity \nearrow

← Harmonics

I4. "Back Door" to QM Let There Be Music



← half-wave
 $n \frac{\lambda}{2} = L \quad \lambda = \frac{2L}{n}$

$$E = \frac{1}{2} m v^2 = \frac{1}{2} \frac{m^2 v^2}{m} = \frac{p^2}{2m}$$

$$p = \frac{h}{\lambda} \Rightarrow E = \frac{1}{2m} \frac{h^2}{\lambda^2} = \frac{h^2}{2m \left(\frac{2L}{n}\right)^2}$$

$$E_n = \frac{n^2 h^2}{8mL^2}$$

The Quantum mechanical result for a particle in a 1D Box!

Also written $E_n = \frac{n^2 (2\pi\hbar)^2}{8mL^2} = \frac{\hbar^2 k^2}{2m}$

$\lambda_3 = \frac{2L}{3}$
 checks out

I5. Schrödinger Equation

We are in search of a differential equation, describes the above waves.

Also called Standing Waves.

$$\psi(x,t) = A \sin(kx) \cos(\omega t)$$

$$E = \frac{p^2}{2m} \quad E = \hbar\omega \quad p = \frac{h}{\lambda} = \hbar \frac{k}{2\pi} = \hbar k$$

$$\hbar\omega = \frac{\hbar^2 k^2}{2m}$$

Can now make $\hbar\omega = \frac{\hbar^2 k^2}{2m}$ work

$$\frac{\partial^2 \psi}{\partial x^2} = -k^2 \psi$$

$$\begin{cases} \frac{d}{dx} \sin kx = k \cos kx \\ \frac{d^2}{dx^2} \sin kx = -k^2 \sin kx \end{cases}$$

$$\frac{\partial \psi}{\partial t} = -\omega A \sin(kx) \sin(\omega t)$$

not equal to $-\omega \psi$

Use $\psi(x,t) = A \sin(kx) e^{-i\omega t}$

Then $\frac{\partial \psi}{\partial t} = -i\omega \psi$

like engineering

Why we need $i = \sqrt{-1}$ in QM. To get $\hbar\omega \psi$

For $\frac{p^2}{2m} + V = E$

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} = i\hbar \frac{\partial \psi}{\partial t}$$

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + V\psi = i\hbar \frac{\partial \psi}{\partial t}$$

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = i\hbar \frac{\partial \psi}{\partial t}$$

Time independent form
 1Dim.